

The Crystal Structure of Palladium n-Propyl Mercaptide

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(Received 17 April 1967 and in revised form 19 December 1967)

The crystal structure of palladium-n-propyl mercaptide has been determined from Patterson and Fourier syntheses using three-dimensional counter data and refined by least squares. The final R index is 0.053 2833 observed reflexions. The space group is $P\bar{1}$ and the cell constants are (at 75°F) $a = 13.766 \pm 0.004 \text{ \AA}$, $b = 10.564 \pm 0.004 \text{ \AA}$, $c = 11.499 \pm 0.004 \text{ \AA}$, $\alpha = 98.36 \pm 0.02^\circ$, $\beta = 111.40 \pm 0.02^\circ$, $\gamma = 102.83 \pm 0.02^\circ$. The molecule exists as a hexamer $\{\text{Pd}(\text{SC}_3\text{H}_7)_2\}_6$ in the crystal. The palladium atoms form a six-membered puckered ring. Each adjacent pair of metal atoms is joined by a double mercaptide bridge which is folded along the sulphur-sulphur axis. Each palladium is surrounded by four sulphur atoms in an approximately square planar configuration.

Introduction

This work forms part of a programme of study of the structures of various metal mercaptides. Results of the structure analysis of mercury mercaptides by Bradley & Kunchur (1964, 1965) and Kunchur (1964) show that the sulphur of the mercapto group acts as a bridge between two mercury atoms. This gives rise to interesting structural arrangements in these compounds. Mercaptides of palladium were prepared by Mann & Purdie (1935) and Hayter & Humieć (1964). The ethyl and phenyl derivatives were found to be highly insoluble and hence linear polymeric structures for these compounds were proposed. The compounds of palladium with n-propyl and higher mercaptans are, however, soluble in organic solvents and can be readily crystallized. Preliminary work on single crystals of palladium n-propylmercaptide showed that the unit cell is triclinic and contains six formula units of this compound (Mann & Purdie, 1935). This is consistent with the results of the molecular weight determination of this compound in solution, which showed the molecular complexity n to have a value approaching six. On the basis of these data, Hayter & Humieć (1964) proposed two possible structures for palladium n-propyl mercaptide, both of which are based on square planar palladium (II).

In one structure palladium atoms are at the corners of a regular octahedron and the bridging mercaptide groups lie above each edge. In the other possibility, the palladium atoms form a six-membered planar ring, each adjacent pair of metals being joined by a double mercaptide bridge, which is folded along the sulphur-sulphur axis. A detailed structure analysis of this compound was undertaken to discover the correct one of these possibilities, and to define the molecular geometry.

Experimental

The crystals of palladium n-propyl mercaptide, kindly supplied by Dr Hayter, were small, well defined orange-red needles along the b -axis. The unit-cell dimensions and space group of this compound were determined

by Mann & Purdie (1935). Their values are: $a = 14.3$, $b = 10.3$, $c = 11.3 \text{ \AA}$, $\alpha = 98^\circ$, $\beta = 113.5^\circ$, $\gamma = 101^\circ$. There are six formula units of this compound in the unit cell. The space group could be $P1$ or $P\bar{1}$, but, the piezoelectric test being negative, they assigned the space group $P\bar{1}$ to the compound.

In the present work, the unit-cell dimensions and the space group were redetermined on the G.E. XRD-6 counter diffractometer using $\text{Cu } K\alpha$ radiation. The results are as follows: palladium n-propyl mercaptide $\{\text{Pd}(\text{S.C}_3\text{H}_7)_2\}_6$. Molecular weight: 1539.5. Triclinic: $a = 13.766 \pm 0.004 \text{ \AA}$, $b = 10.564 \pm 0.004 \text{ \AA}$, $c = 11.499 \pm 0.004 \text{ \AA}$, $\alpha = 98.36 \pm 0.02^\circ$, $\beta = 111.40 \pm 0.02^\circ$, $\gamma = 102.83 \pm 0.02^\circ$. The dimensions were obtained using high order axial reflexions under the 'fine conditions' of the diffractometer, at 75°F. The volume of the unit cell is 1470.2 \AA^3 . The absorption coefficient for X-rays, $\lambda = 1.5404 \text{ \AA}$, $\mu = 188.0 \text{ cm}^{-1}$. The density = 1.73 g.cm^{-3} , calculated on the basis of six formula units of the monomeric compound in the unit cell, agrees approximately with the experimental value of 1.75 g.cm^{-3} . Total number of electrons in the unit cell $F(000) = 768$.

The space group $P\bar{1}$ was assumed to be correct on the basis of the work of the previous authors. The crystals employed for the unit-cell measurements and data collection were needles of dimensions $0.06 \times 0.06 \times 0.05 \text{ mm}$. For the linear absorption coefficient $\mu = 188.0 \text{ cm}^{-1}$, the value for μr is 0.56 . φ -absorption corrections were applied. Data were collected on the G.E. XRD-6 diffractometer with scintillation counter using $\text{Cu } K\alpha$ radiation and a balanced pair of cobalt and nickel filters. The stationary-crystal stationary-counter technique was used. All 2929 possible independent reflexions within a range $0^\circ < 2\theta < 100^\circ$ were examined. Of these, 96 were too weak to be measured. α -splitting and the usual $1/L_p$ corrections were applied and the raw intensities were converted into F^2 and F .

Structure determination

A three-dimensional 'unsharpened' Patterson synthesis was computed. In the space group $P\bar{1}$, there are eight centres of symmetry and since there are six palladium

atoms in the unit cell, the job of fixing the palladium positions was rather complicated. The structure was solved by use of Buerger's 'image seeking' method. A palladium-palladium single weight peak was located and a line was drawn from the origin to the center of the peak. This line was used as an image to locate the remaining palladium atoms. No attempt was made to locate the sulphur atoms at this stage.

Three cycles of structure-factor least squares were carried out for the six palladium atoms. This gave an agreement index $R=0.31$. A palladium-phased three-dimensional electron density Fourier series was then computed. This yielded the positions of all the twelve sulphur and two carbon atoms. The agreement index R at this stage was 0.21. A second electron density difference Fourier series was computed, using the coefficients obtained by subtracting the contributions of all palladium and sulphur atoms from the F_{obs} . This gave the coordinates of all but two carbon atoms. The agreement index R for all the atoms located up to this stage was 0.16. Another electron density difference Fourier series was computed; this gave the remaining two carbon atoms. The structure, now essentially determined, was refined by a series of least-squares analyses of the three-dimensional data. Four cycles, with isotropic atomic temperature factors, were carried out on the IBM 7040 computer, using the program written by Okaya and modified by Kärtha & Harris. The agreement index R for all observed reflexions fell to 0.11. Another four cycles of refinement with individual anisotropic temperature factors and a block-diagonal approximation, reduced the R to 0.053. A $1/f$ weighting with f corresponding to the contribution of sulphur

was adopted in the final stages of the least-squares analysis. This weighting is approximately equivalent to differential synthesis weighting. Atomic scattering factors for palladium, sulphur, and carbon were taken from the *International Tables for Crystallography*, Vol. III (1962). The anomalous corrections $\Delta f'$ and $\Delta f''$ for palladium with $\text{Cu } K\alpha$ were applied. The atomic shifts in the final cycle were much smaller than the standard deviations of the atomic positions, and so the refinement was considered to be complete. An electron density difference Fourier series, computed using the final coordinates of all the atoms except the hydrogens, was free from details of any significance. No attempt was made to locate the hydrogen atoms; the calculations described above do not take into account the contributions of hydrogen atoms to the structure factors. The final agreement index R was 0.053 for all observed reflexions and 0.058 if all unobserved reflexions were included. The final atomic and thermal parameters as well as the F_{obs} and F_{calc} are those of the final least-squares cycle. The atomic and thermal parameters are listed in Tables 1 and 7 while the F_{obs} and F_{calc} are listed in Table 2.

Discussion and description of the structure

The various bond lengths, bond angles and their estimated standard deviations are shown in Table 3. The standard deviations in the atomic parameters were obtained by inverting the matrix in the final least-squares cycle. The labelling scheme used to describe the palladium n-propyl mercaptide molecule can be seen in Fig. 1. For every atom in the unit cell, there is another atom

Table 1. Fractional final coordinates and their standard deviations (\AA)

	x/a	y/b	z/c	σ_x	σ_y	σ_z
Pd(1)	0.16796	0.01108	0.27509	0.00070	0.00138	0.00089
Pd(2)	0.01487	0.25220	0.88932	0.00072	0.00140	0.00090
Pd(3)	0.18416	0.26365	0.16557	0.00070	0.00138	0.00088
S(1)	0.0583	0.3800	0.0945	0.0026	0.0049	0.0032
S(2)	0.3049	0.1390	0.2302	0.0024	0.0049	0.0032
S(3)	0.1538	0.2218	0.3455	0.0028	0.0051	0.0033
S(4)	0.8245	0.1997	0.8081	0.0026	0.0051	0.0032
S(5)	0.9739	0.1179	0.6877	0.0026	0.0052	0.0032
S(6)	0.2018	0.2844	0.9750	0.0026	0.0054	0.0033
C(1)	0.948	0.327	0.1469	0.013	0.0251	0.0163
C(2)	0.967	0.422	0.270	0.013	0.025	0.015
C(3)	0.882	0.377	0.317	0.020	0.039	0.025
C(4)	0.307	0.038	0.087	0.010	0.023	0.013
C(5)	0.420	0.111	0.079	0.016	0.033	0.019
C(6)	0.517	0.099	0.182	0.016	0.043	0.022
C(7)	0.285	0.308	0.480	0.012	0.023	0.014
C(8)	0.308	0.457	0.509	0.021	0.029	0.023
C(9)	0.581	0.477	0.373	0.022	0.038	0.023
C(10)	0.782	0.269	0.663	0.011	0.023	0.014
C(11)	0.668	0.190	0.562	0.012	0.026	0.015
C(12)	0.582	0.198	0.608	0.016	0.040	0.028
C(13)	0.080	0.034	0.702	0.012	0.022	0.015
C(14)	0.162	0.105	0.663	0.010	0.025	0.013
C(15)	0.245	0.031	0.671	0.013	0.036	0.020
C(16)	0.259	0.467	0.996	0.012	0.028	0.017
C(17)	0.578	0.341	0.937	0.020	0.040	0.036
C(18)	0.624	0.484	0.924	0.019	0.036	0.030

Table 2. Observed and calculated structure factors
The unobserved reflections are marked by an asterisk.

H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL										
0	1	0	1838	1877	0	1	6	932	942	-1	7	-1	864	854	*1	4	4	81	71	-1	3	-7	919	960	-2	2	1	1365	1181					
0	2	0	2243	2417	0	2	6	1756	1682	1	7	-1	1702	1757	-1	4	4	148	149	1	3	-7	919	922	-2	2	-1	2932	2645					
0	3	0	1148	1155	0	2	-6	932	932	1	8	-1	1135	1107	-1	4	-4	445	544	1	4	-7	621	662	2	2	1	1486	1436					
0	4	0	3446	3442	0	3	-6	905	969	1	8	1	891	1579	1	5	-4	959	906	-1	4	-7	1067	1027	-1	4	7	1013	981	-2	3	-1	3108	3185
0	5	0	324	468	0	4	6	581	506	-1	8	1	1486	1517	-1	5	4	1067	1027	-1	4	7	1013	981	-2	3	1	513	446					
0	6	0	824	831	0	4	-6	189	170	-1	8	-1	351	306	-1	5	-4	1310	1295	-1	4	-7	324	359	-2	3	-1	378	251					
0	7	0	1081	1128	0	5	-6	689	677	1	9	1	527	552	1	5	4	324	339	-1	5	7	466	432	2	3	1	1324	1341					
0	8	0	1783	1796	0	5	6	864	840	-1	9	1	202	125	-1	6	-4	1067	1009	-1	5	7	1675	1687	2	4	1	1473	1452					
#	0	9	0	94	62	0	6	-6	1297	1260	-1	9	-1	405	426	1	6	-4	3081	3065	1	5	7	824	847	-2	4	-1	1892	1836				
0	0	1	2365	2489	0	6	6	243	168	1	9	-1	446	462	*1	6	4	94	101	-1	6	-7	608	613	-2	4	-1	1081	1189					
0	1	-1	391	345	0	7	6	932	937	-1	10	-1	810	842	-1	6	4	797	825	1	6	7	729	753	2	4	-1	1081	1189					
0	1	1	1675	1547	0	8	-6	527	527	-1	0	2	891	727	1	7	4	621	600	-1	6	7	243	211	2	5	-1	175	132					
0	2	1	1337	1384	0	9	-6	243	199	1	0	2	3013	2867	1	7	4	243	198	1	7	-7	378	352	-2	5	1	2905	3010					
0	2	-2	3121	3126	0	0	7	837	799	1	1	-2	2243	2174	-1	7	-4	1743	1714	-1	7	-7	229	270	-2	5	-1	1486	1616					
0	3	1	1283	1186	0	1	7	202	192	-1	1	2	310	394	1	8	-4	932	939	1	9	-7	959	898	2	6	1	391	417					
0	4	1	1648	1652	0	3	7	2162	2128	1	2	1	2351	1254	-1	8	-4	554	523	1	0	8	297	273	-2	6	-1	405	499					
0	5	-1	297	315	0	3	-7	756	730	-1	2	2	1527	1441	-1	9	-4	459	453	-1	0	8	756	743	2	6	-1	716	727					
0	5	1	837	848	0	4	7	689	670	-1	2	-2	2405	2172	1	9	-4	459	461	-1	1	8	864	870	2	7	-1	1175	1322					
0	6	-1	1027	1073	0	4	-7	1337	1322	1	2	2	986	999	-1	10	-4	648	607	1	1	8	1094	999	-2	7	1	1162	1154					
0	6	1	1121	1172	0	5	7	675	689	1	3	2	1581	1601	-1	0	5	2243	2195	1	1	8	186	173	-2	7	-1	527	483					
0	7	-1	324	359	0	5	-7	1459	1418	-1	3	3	2054	2012	1	0	5	1540	1590	1	2	8	756	747	2	7	1	783	766					
0	7	1	216	109	0	6	-7	135	104	-1	3	-2	1202	1248	1	1	-5	189	220	1	2	8	256	211	-2	8	-1	337	424					
0	8	-1	1337	1415	*0	7	-7	94	95	1	3	-2	1148	1196	1	1	5	1310	1314	-1	2	8	135	82	2	8	-1	243	264					
0	9	1	932	983	0	8	-7	540	498	1	4	-2	202	233	-1	1	5	1162	1297	*1	2	-8	81	88	-2	8	1	1851	1783					
0	10	-1	337	343	0	9	-7	527	516	1	4	2	729	657	-1	1	5	1027	918	-1	3	8	1283	1291	-2	9	1	1432	1448					
0	0	2	2716	2645	0	0	8	175	140	-1	4	2	1243	1137	1	2	5	1364	1337	-1	3	-8	756	769	-2	9	-1	824	819					
0	1	-2	2878	2867	0	-1	8	500	479	-1	4	-2	527	513	-1	2	5	1838	1735	1	3	-8	243	284	2	9	-1	540	651					
0	1	2	3297	3025	0	1	8	635	617	1	5	-2	1756	1753	-1	2	-5	1959	1956	1	4	-8	2027	2035	-2	10	1	148	90					
0	2	2	1229	1140	0	3	8	837	835	-1	5	2	202	217	1	2	-5	3013	2901	1	4	8	253	234	-2	10	-1	1040	1045					
0	2	-2	621	601	0	3	-8	473	499	-1	5	-2	1500	1423	1	3	5	486	409	-1	4	8	716	770	2	0	-2	1067	1067					
0	3	2	1756	1762	0	4	-8	878	887	1	5	2	1202	1211	-1	3	5	946	861	-1	4	-8	297	320	2	0	2	891	906					
0	3	-2	824	871	0	5	-8	1202	1253	-1	6	-2	337	372	-1	3	-5	702	741	1	5	8	1351	1291	2	1	2	2757	2663					
0	4	-2	1513	1504	0	6	-8	135	137	1	6	-2	1013	1040	1	3	-5	986	946	-1	5	8	283	269	-2	1	2	2838	2637					
0	4	-4	216	148	0	7	-8	189	146	1	6	2	621	558	1	4	-5	1540	1475	1	5	8	270	322	2	1	-2	391	339					
0	5	-2	2081	1966	0	8	-8	567	605	-1	6	2	797	770	1	4	5	121	89	1	6	-8	621	591	2	1	-2	1310	1212					
0	5	2	297	282	0	0	9	135	139	1	7	2	621	596	-1	4	5	716	706	1	6	-8	432	389	2	2	-2	1027	930					
0	6	-2	1229	1211	0	-1	9	189	132	-1	7	2	283	304	-1	4	-5	500	409	-1	7	-8	459	439	-2	2	2	1959	1732					
#	6	2	81	47	0	1	9	540	506	1	8	-2	486	533	1	5	-5	418	413	-1	8	-8	135	117	-2	2	-2	973	959					
0	7	-2	1067	1050	0	2	9	810	876	-1	8	2	459	429	-1	5	5	283	210	1	8	-8	1027	992	2	2	2	391	478					
0	7	2	459	493	0	-2	9	229	231	-1	8	-2	756	761	-1	5	5	932	918	1	0	9	310	311	2	3	-2	2635	2644					
0	8	2	1027	993	0	3	9	486	480	-1	9	2	1067	1038	1	5	5	648	685	-1	1	9	297	259	-2	3	2	1594	1585					
0	8	-2	405	357	0	3	-9	756	822	-1	9	-2	770	775	-1	6	-5	1594	1552	1	1	9	243	174	-2	3	-2	3013	2840					
0	9	-2	567	559	0	4	-9	797	828	1	9	-2	445	432	1	6	-5	2878	2853	1	1	9	459	488	2	3	2	1540	1534					
0	10	-2	513	558	0	6	-9	432	471	1	10	-2	337	361	-1	6	5	432	420	1	2	9	810	881	-2	4	2	297	325					
0	0	3	337	412	0	7	-9	391	433	-1	0	3	1283	1306	*1	7	5	94	52	-1	2	9	310	260	-2	4	-2	189	215					
0	1	-3	445	353	0	0	10	202	241	1	0	3	1081	1025	-1	7	5	446	439	-1	2	9	540	542	2	4	-2	864	898					
0	1	3	2311	2237	0	-1	10	615	755	1	1	-3	175	194	-1	7	5	959	912	1	2	-9	378	334	2	5	-2	135	132					
0	2	2	2982	2822	0	1	10	635	674	1	1	3	3284	3273	1	7	5	473	489	*1	3	9	94	126	-2	5	2	1513	1485					
0	2	-3	2432	2448	*0	-2	10	94	130	-1	3	-1	312	1105	-1	6	-5	1207	1002	1	1	6	783	731	-2	5	-2	1919	1967					
0	3	3	1648	1561	0	4	-10	459	453	1	2	3	1297	1403	-1	9	-5	418	420	1	4	-9	445	441	2	6	2	256	249					
0	4	-3	554	531	0	5	-10	621	678	1	2	-3	1727	1543	1	9	-5	337	341	-1	4	-9	372	372	-2	6	2	1054	1078					
0	4	-3	1013	962	0	6	0	4284	4544	-1	2	-3	3040	2948	-1	10	-5	189	231	1	5	5	635	685	-2	6	2	986	984					
0	5	-3	675	638	-1	1	0	3																										

Table 2 (cont.)

H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL						
2	8	3	135	157	-2	8	-6	364	363	3	6	0	1297	1216	-3	4	3	1716	1685	-3	3	-6	270	338	4	1	0	229	190	
-2	8	3	756	722	2	8	-6	689	649	-3	7	0	905	892	3	5	-3	1040	1035	-3	4	-6	1324	1316	-4	2	0	1594	1591	
-2	9	3	594	584	2	9	-6	608	585	3	7	0	324	288	3	5	3	324	405	3	4	-6	216	127	4	2	0	1243	1312	
-2	9	-3	135	148	-2	9	-6	1094	1074	-3	8	0	243	247	-3	5	3	243	174	3	4	6	581	653	-4	3	0	919	890	
2	9	-3	243	279	-2	0	7	297	259	*-	3	8	0	94	63	-3	5	-3	1243	1261	-3	4	6	702	647	4	3	0	1338	1158
-2	10	-3	891	821	2	0	7	662	587	-3	9	0	256	169	-3	6	-3	459	469	3	5	-6	1365	1325	-4	4	0	1081	1119	
-2	0	4	2838	2719	2	1	7	621	697	*	-3	10	0	94	680	3	6	-3	162	92	-3	5	6	472	439	4	4	0	1013	989
2	0	4	1000	1498	-2	1	7	783	778	3	0	1	1121	1194	3	6	3	689	691	-3	5	-6	864	867	-4	5	0	2378	2406	
2	1	4	94	91	-2	1	7	229	191	3	0	-1	500	433	-3	6	3	783	777	-3	6	-6	1446	1419	4	5	0	229	161	
-2	1	4	1581	1644	-2	1	7	189	123	3	1	1	1716	1890	-3	7	-3	797	775	3	6	-6	919	912	-4	6	0	675	677	
-2	1	4	1027	918	2	2	-7	1567	1510	-3	1	1	2500	2387	3	7	-3	689	699	-3	6	1	1108	1038	4	6	0	1351	1287	
2	1	-4	1878	1902	2	-2	7	1121	1115	-3	1	-1	1716	1623	3	7	3	459	455	-3	7	6	581	556	-4	7	0	459	507	
2	2	-4	756	758	-2	2	7	1229	1218	3	1	-1	1797	1747	-3	7	3	797	706	-3	7	6	135	81	*	4	7	0	94	76
-2	2	4	2689	2493	2	2	7	270	257	3	2	1	297	340	-3	8	-3	270	232	3	7	-6	500	481	-4	8	0	135	155	
-2	2	-4	1581	1570	2	3	-7	1838	1770	-3	2	1	1135	1032	3	8	-3	135	86	-3	8	-6	135	122	4	8	0	662	676	
2	2	4	229	225	-2	3	7	770	703	-3	2	-1	418	344	-3	8	3	135	237	3	9	-6	1162	1134	-4	9	0	527	549	
2	3	-4	1256	1180	-2	3	-7	797	796	3	2	-1	602	667	3	9	-3	608	607	-3	9	1	1013	970	-4	10	0	581	601	
-2	3	4	770	855	2	3	7	464	484	3	3	-1	1148	1264	-3	9	3	283	300	3	0	7	756	693	4	0	1	3419	3788	
-2	3	-4	486	423	2	4	7	594	627	3	3	1	1446	1543	-3	9	-3	446	384	-3	0	7	932	844	4	0	-1	1540	1452	
2	3	4	716	689	-2	4	7	189	131	3	3	1	1162	1203	3	0	4	1824	1773	3	1	7	513	413	4	1	-1	1932	1928	
2	4	4	797	824	-2	4	-7	527	573	-3	3	-1	2527	2518	-3	0	4	148	157	3	-1	7	783	763	4	-1	1	283	517	
-2	4	4	797	797	2	4	-7	567	553	3	4	-1	797	763	3	1	4	202	269	-3	1	7	797	791	-4	1	1	189	220	
-2	4	-4	527	480	2	5	-7	419	372	3	4	1	1702	1852	-3	1	-1	420	1770	*	3	1	7	81	80	4	1	1	1905	1990
2	4	-4	1540	1534	-2	5	7	459	412	-3	1	4	324	335	-3	1	4	1581	1653	3	2	7	432	503	4	2	1	1567	1660	
2	5	-4	878	906	-2	5	-7	878	960	3	5	-1	851	898	3	1	-4	202	141	3	-2	7	432	438	-4	2	1	310	384	
-2	5	4	324	279	2	6	-7	567	581	3	5	1	1878	2038	3	2	4	486	477	-3	2	7	405	386	-4	2	-1	1635	1687	
-2	5	-4	1000	1009	-2	6	7	567	638	-3	5	1	2635	2715	-3	2	4	175	114	3	2	-7	1054	1020	4	2	-1	662	621	
2	5	4	554	553	-2	6	-7	500	501	-3	5	-1	1932	1893	-3	2	-4	756	669	3	3	-7	2067	1991	-4	3	-1	3351	3294	
2	6	4	337	395	-2	7	-7	554	476	-3	6	-1	108	78	3	2	-4	2189	2133	3	3	3	283	243	4	3	-1	1310	1262	
-2	6	4	1256	1252	*	-2	7	7	94	91	3	6	-1	175	168	3	3	-6	2067	2013	-3	3	7	581	547	4	3	1	67	54
-2	6	-4	1013	981	-2	8	-7	243	240	3	6	1	567	605	3	3	4	567	551	-3	3	-7	905	1011	-4	3	1	1500	1603	
2	6	-4	162	127	2	8	-7	445	503	-3	6	1	297	287	-3	3	4	986	890	-3	4	-7	635	738	4	4	-1	1513	1450	
2	7	4	418	445	-2	0	8	1270	1324	-3	7	-1	621	639	-3	4	-7	1702	1709	3	4	-7	608	590	4	4	1	3486	3668	
2	7	-6	1189	1207	-2	0	8	378	291	3	7	-1	391	391	-3	4	-4	567	540	-3	4	7	554	533	-4	4	1	2365	2501	
-2	7	4	540	548	2	1	8	946	847	3	7	1	419	422	-3	4	-4	1121	1128	3	5	-7	216	216	-4	4	-1	3405	3349	
2	8	-4	229	220	-2	-1	8	662	640	-3	7	1	1662	1719	3	4	-4	1088	1088	-3	5	7	743	650	4	5	-1	1081	1054	
2	8	-4	297	347	-2	1	8	283	286	-3	8	-1	135	100	-3	4	4	1892	1895	-3	5	-7	324	353	4	5	1	1716	1770	
-2	8	4	513	515	-2	-1	8	662	675	*	3	8	-1	94	80	3	5	-4	608	606	3	6	-7	297	174	-4	5	1	405	405
2	9	-4	459	481	2	2	-8	594	602	3	8	1	729	833	3	5	4	554	603	-3	6	7	581	581	-4	5	-1	216	132	
-2	10	4	337	352	-2	2	8	567	513	3	9	-1	513	504	-3	5	-4	189	171	-3	7	7	513	562	-4	6	1	554	587	
-2	0	5	2946	2873	2	2	8	986	935	-3	9	1	1324	1328	-3	6	-4	608	599	3	8	-7	540	503	4	6	-1	1054	1001	
2	0	5	946	940	2	3	-8	270	239	-3	9	-1	864	799	3	6	-4	973	950	-3	8	-7	554	533	4	7	1	94	70	
2	1	5	1027	992	-2	3	8	743	729	*	-3	10	1	94	104	3	6	4	148	232	3	0	8	608	551	-4	7	1	1040	1146
-2	1	5	797	802	-2	3	-8	635	592	-3	10	-1	554	551	-3	6	4	243	231	-3	0	8	581	575	-4	7	-1	1094	1089	
2	1	-5	2189	2195	2	3	8	243	198	3	0	2	2878	2655	*	3	7	4	81	567	3	1	8	310	264	4	7	-1	148	82
-2	2	-5	3905	3860	-2	4	8	1148	1180	3	0	-2	2405	2221	3	7	-6	824	833	3	-1	8	337	317	*	4	8	-1	94	81
-2	2	5	3797	3724	-2	4	-8	189	154	3	2	2	2784	2678	-3	7	4	378	367	-3	1	8	175	99	4	8	1	946	915	
2	2	-5	283	247	2	4	-8	1540	1570	-3	1	2	1865	1880	-3	8	4	689	692	3	2	-8	878	844	-4	8	-1	675	630	
-2	2	5	554	577	2	5	-8	135	118	3	1	-2	1466	1286	-3	8	-4	608	564	-3	2	8	946	922	-4	9	-1	337	312	
-2	3	5	2040	1949	2	5	-8	229	232	3	2	2	1148	1199	-3	9	4	310	331	3	2	-8	1513	1502	-4	9	1	418	395	
-2	3	5	1756	1717	-2	6	-8	662	679	-3	2	2	1392	1307	-3	9	-4	310	321	3	3	-8	648	610	-4	10	-1	216	129	
2	3	5	810	790	-2	6	-8	973	900	-3	2	-2	283	358	3	9	-4	202	169	3	3	-8	189	156	*	4	10	1	94	58
2	4	5	364	339	2	7	-8	432	359	3	2	0	1599	2067	-3	10	-4	770	754	-3	8	-8	405	442						

Table 2 (cont.)

H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL					
4	3	3	1337	1287	4	2	-6	1675	1666	5	4	0	2784	2633	-5	5	3	2202	2229	-5	7	6	1797	1746	-6	2	1	1986	2049
-4	3	3	837	906	-4	3	-6	216	113	-5	5	0	486	541	5	6	-3	1662	1741	* -5	7	-6	94	69	-6	2	-1	1594	1552
4	4	-3	229	216	4	3	-6	175	245	5	5	0	243	170	5	6	3	540	533	5	7	-6	648	651	6	2	-1	932	954
4	4	3	1270	1287	4	3	6	432	503	-5	6	0	851	907	-5	6	3	1648	1658	-5	8	6	662	650	6	3	1	689	537
-4	4	3	405	423	-4	3	6	2419	2323	5	6	0	1013	1012	-5	6	-3	662	579	-5	8	-6	324	350	-6	3	1	567	651
-4	4	-3	770	775	4	4	-6	1108	1086	-5	7	0	608	598	5	7	-3	608	648	5	8	-6	135	144	-6	3	-1	1135	1079
4	5	-3	513	548	4	4	6	351	471	5	7	0	621	550	-5	7	3	297	241	5	0	7	594	554	5	3	-1	1067	1007
4	5	3	783	868	-4	4	6	175	172	-5	8	0	918	947	-5	7	-3	135	127	-5	0	7	1297	1274	6	4	-1	1175	1156
-4	5	3	2094	2063	-4	4	-6	1135	1226	-5	9	0	378	332	5	8	-3	473	463	-5	1	7	364	286	6	4	1	148	80
-4	5	-3	1770	1757	4	5	-6	986	961	-5	10	0	554	605	-5	8	3	472	428	5	1	-7	2108	2064	-6	4	1	878	983
4	6	3	202	205	-4	5	6	905	851	5	0	-1	729	699	-5	8	-3	594	552	5	1	7	500	450	-6	4	-1	2175	2083
-4	6	3	243	262	4	6	-6	310	281	5	0	1	3432	3720	-5	9	-3	148	175	5	-1	7	486	445	6	5	-1	473	415
-4	6	-3	851	810	-4	6	6	1148	1089	-5	1	1	324	337	-5	9	3	959	933	5	2	-7	108	77	6	5	1	1094	905
4	6	-3	1297	1359	-4	6	-6	783	757	5	1	-1	3013	2873	-5	10	-3	256	247	5	-2	7	418	403	-6	5	1	621	589
-4	7	3	770	745	4	7	-6	378	379	5	1	1	216	595	5	0	4	1148	1107	-5	2	7	648	607	-6	5	-1	1148	1139
-4	7	-3	256	210	-4	7	6	1986	1913	5	-1	1	567	630	5	0	-4	878	987	5	3	-7	567	613	6	6	1	310	237
-4	7	-3	1175	1254	4	8	-6	689	672	5	2	-1	2446	2408	-5	1	4	1094	1198	5	3	-7	310	231	-6	6	-1	878	823
* -4	8	-3	94	56	-4	8	-6	783	741	-5	2	1	1135	1135	5	1	-4	1432	1534	-5	3	7	1455	1390	6	6	-1	513	494
* -4	8	-3	378	379	4	0	7	270	224	-5	2	-1	2716	2709	5	1	4	1554	1517	-5	4	-7	1027	989	-6	6	1	1446	1528
* -4	8	3	94	81	-4	0	7	283	276	5	2	1	966	1166	-5	1	4	337	333	-5	4	7	405	331	6	7	-1	892	850
-4	9	-3	432	406	4	1	-7	1459	1452	-5	3	1	1027	960	5	2	-4	1175	1249	-5	5	7	1000	877	-6	7	1	743	810
-4	9	3	405	375	-4	1	7	864	849	5	3	-1	824	795	5	2	-4	2716	2666	-5	5	-7	648	644	-6	7	-1	243	137
-4	9	-3	297	279	-4	1	7	1310	1315	5	3	1	851	940	-5	2	4	378	331	5	5	-7	743	762	-6	8	-1	1067	1063
-4	10	-3	770	741	4	1	7	243	234	-5	3	1	364	300	5	2	4	1324	1340	-5	6	7	135	162	* -6	8	1	94	105
-4	0	4	1338	1352	4	2	7	662	610	5	4	1	905	2394	-5	3	-4	1675	1751	5	6	-7	324	315	-6	9	1	689	656
-4	1	-4	3027	2970	-4	2	7	1000	934	-5	4	1	2013	1979	5	3	-4	1608	1641	5	7	-7	243	240	-6	9	-1	554	589
-4	1	-4	513	501	4	2	-7	162	139	-5	5	-1	1513	1461	-5	3	4	1351	1356	-5	7	-7	500	482	-6	10	1	675	700
-4	1	4	162	96	-4	3	-7	919	1007	5	5	0	837	830	5	4	4	459	443	5	8	-7	1175	1149	6	0	2	1473	1596
-4	1	4	810	701	4	3	-7	1081	1118	5	5	1	81	256	-5	4	4	1067	1127	-5	0	8	783	764	-6	0	-2	2270	2269
-4	2	4	1365	1317	-4	3	7	513	491	-5	5	1	2121	2161	5	4	-4	310	386	-5	1	8	1094	1107	-6	1	2	243	371
-4	2	4	310	320	4	4	-7	310	262	5	6	-1	891	928	5	4	-4	1797	1781	5	1	-8	364	380	6	1	-2	2729	2753
-4	2	-4	1554	1598	-4	4	7	432	365	5	6	1	648	677	-5	5	-4	175	188	5	1	-8	932	922	6	1	2	554	548
-4	2	-4	2297	2199	-4	4	-7	297	333	5	6	1	973	996	5	5	-4	675	622	5	2	-8	418	424	-6	1	2	756	783
-4	3	-4	364	392	-4	5	7	959	894	-5	6	-1	297	332	5	5	4	986	1056	-5	2	8	729	692	6	2	-2	851	938
-4	3	-4	459	388	-4	5	-7	189	158	5	7	1	472	521	-5	5	4	716	677	5	2	-8	567	534	6	2	2	581	615
-4	3	4	189	234	4	5	-7	594	525	-5	7	1	756	824	5	6	-4	1297	1379	5	3	-8	824	843	-6	2	2	202	176
-4	3	4	1554	1547	-4	6	-7	391	369	-5	7	1	824	821	-5	6	4	256	301	5	3	-8	202	250	-6	2	-2	243	226
-4	4	-4	851	798	-4	7	7	135	79	5	7	-1	256	205	-5	6	-4	1783	1769	-5	3	8	189	191	6	3	-2	256	249
-4	4	4	418	504	-4	7	-7	648	655	5	8	-1	256	274	-5	7	-4	1067	1070	5	4	8	310	252	6	3	2	878	915
-4	4	4	1148	1171	4	7	-7	310	320	-5	8	1	648	669	5	7	-4	500	543	-5	4	8	243	235	-6	3	2	905	967
-4	4	-4	1270	1275	4	8	-7	391	419	-5	8	-1	1121	1186	-5	7	4	919	931	-5	4	8	662	746	-6	3	-2	189	108
-4	5	-4	986	997	-4	8	-7	391	428	-5	9	-1	337	263	-5	8	-4	1216	1107	5	5	8	459	446	-6	4	-2	581	520
-4	5	4	432	448	4	0	8	864	965	-5	9	1	473	529	5	8	-4	351	401	-5	5	8	486	450	6	4	-2	216	233
-4	5	4	932	950	-4	0	8	851	824	-5	10	0	378	373	-5	8	4	554	541	-5	5	8	878	957	6	4	2	675	662
-4	5	-4	202	197	4	1	-8	973	956	-5	10	1	297	221	5	9	-4	148	102	-5	6	8	202	173	-6	4	2	1297	1357
-4	6	4	892	937	-4	3	-8	608	749	5	0	-2	1554	1484	-5	9	-4	202	1247	-5	1	9	189	136	6	6	-2	851	869
-4	6	-4	946	948	-4	3	8	513	492	-5	2	2	1297	1221	-5	3	5	378	375	-5	0	10	702	670	-6	9	-2	378	384
-4	6	-4	202	200	4	4	-8	175	167	5	2	2	1405	1383	-5	5	2	5	243	215	-6	7	-2	554	529				
-4	9	-4	513	554	-4	4	8	337	318	-5	2	-2	1527	1500	-5	4	-5	283	202	5	5	-9	756	708	-6	7	-2	418	376
-4	10	-4	243	146	* -4	5	-8	94	147	5	3	2	919	841	-5	3	-5	1135	1204	-5	3	10	202	163	-6	8	-2	297	334
-4	0	5	1392	1400	-4	5	8	445	395	-5	3	2	162	154	5	3	-5	148	154	-5	7	9	337	351	-6	8	2	824	844
-4	0	5	500	503	-4	5	8	648	709	5	4	-2	1162	1218	5	3	5	378	375	-5	0	10	702	670	-6	9	-2	378	384
-4	1	-5	1013	945	4	6	-8	135	131	5	4	2	729	627	-5	3	5	391	398	-5	1	10	865	902	-6	9	-2	378	384
-4	1	-5																											

Table 2 (cont.)

H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	H	K	L	F OBS	F CAL	
6	-1	4	702	700	-6	6	7	310	303	-7	3	-2	1000	1023	7	5	-5	1216	1202	8	1	-1	243	297	
6	2	-4	3094	3157	-6	6	-7	202	249	7	3	-2	648	630	-7	6	5	810	889	8	-1	-1	3000	2885	
6	2	4	540	494	* 6	7	-7	94	135	7	3	2	648	819	-7	6	-5	1081	1204	8	-1	1	1905	1935	
6	-2	4	1189	1124	-6	7	7	1351	1325	-7	3	2	1216	1202	7	6	-5	1297	1359	8	2	-1	973	887	
-6	2	4	2878	3020	-6	0	8	662	691	7	4	-2	202	216	-7	7	5	351	334	-8	2	1	1121	1163	
6	3	-4	675	690	6	1	-8	1148	1153	-7	4	2	459	372	-7	7	-5	657	658	8	2	1	513	439	
* -6	3	4	94	92	-6	1	8	121	138	-7	4	-2	121	131	-7	8	-5	148	82	8	-2	1	283	353	
-6	3	4	175	240	* -6	2	8	94	68	7	4	-2	189	113	-7	8	5	540	537	-8	3	1	756	761	
-6	3	-4	1216	1318	6	2	-8	648	681	-7	5	2	121	126	7	0	-6	973	904	-8	3	-1	175	222	
-6	4	-4	310	273	-6	3	8	446	440	-7	5	-2	1432	1342	7	0	6	297	251	8	3	-1	675	661	
6	4	-4	2135	2132	6	4	-8	1202	1228	7	5	-2	729	733	7	-1	6	851	802	8	3	1	973	830	
6	4	4	500	406	-6	4	527	479	7	5	2	418	583	-7	1	6	1594	1664	8	4	1	446	334		
-6	4	4	1283	1314	-6	5	8	527	483	-7	6	-2	554	506	7	1	-6	540	515	* -8	4	-1	94	118	
-6	5	-4	189	185	6	5	-8	716	726	-7	6	2	364	359	7	2	-6	459	421	-8	4	-1	567	536	
6	5	-4	648	693	-6	6	8	243	243	7	6	-2	810	732	7	-2	6	527	560	-8	4	1	1162	1202	
-6	5	4	729	752	6	6	-8	283	274	7	7	-2	297	267	-7	2	6	1310	1296	8	5	-1	297	241	
6	6	4	1162	1206	6	7	-8	243	135	-7	7	2	1000	990	7	-3	6	702	656	-8	5	1	2013	2019	
-6	6	-4	1756	1788	-6	0	9	1094	980	-7	7	-2	283	235	7	3	-6	851	659	-8	5	-1	1189	1116	
6	6	-4	2202	2262	6	1	-9	189	214	-7	8	-2	540	570	-7	3	6	1392	1410	-8	6	1	270	256	
-6	7	-4	1986	1978	-6	1	9	878	841	-7	8	2	824	812	7	4	-6	892	921	-8	6	-1	405	405	
6	7	-4	594	653	-6	2	9	946	901	-7	9	2	675	660	-7	4	6	635	597	8	6	-1	946	800	
-6	7	4	243	185	6	2	-9	135	99	-7	9	-2	635	648	-7	4	-6	364	399	-8	7	-1	202	219	
-6	8	-4	513	492	6	3	-9	851	835	7	0	-3	1405	1518	7	5	-6	135	178	-8	7	1	944	969	
6	8	-4	148	170	-6	3	9	621	599	7	0	3	1324	1199	-7	5	6	297	256	-8	8	1	1040	1047	
* -6	8	4	94	124	6	4	-9	500	491	-7	1	3	919	958	-7	5	-6	837	884	-8	8	-1	986	918	
-6	9	4	418	378	-6	4	9	135	107	-7	1	-3	364	305	-7	6	6	554	551	-8	9	1	2013	1985	
* -6	9	-4	608	655	-6	5	9	135	177	7	1	3	2243	2031	7	6	-6	351	333	-8	9	1	1594	1614	
* -6	0	5	94	77	6	5	-9	310	257	-7	1	-3	851	786	7	7	-6	878	912	8	0	-2	378	333	
6	0	-5	1635	1617	6	6	-9	540	548	7	2	3	216	158	-7	7	6	689	676	8	0	2	1324	1358	
-6	1	5	378	315	-6	0	10	635	597	7	2	-3	297	312	-7	7	-6	824	821	8	1	2	1702	1823	
6	1	-5	1216	1245	6	1	-10	608	580	-7	2	3	581	634	7	0	-7	204	804	8	-1	2	337	493	
-6	1	-5	810	769	-6	1	10	689	715	-7	2	3	310	299	7	1	-7	270	245	8	1	-2	1256	1238	
6	2	-5	108	135	-6	2	10	310	287	-7	3	3	189	115	-7	1	7	459	523	-8	1	-2	2027	1906	
6	2	5	729	816	6	2	-10	621	622	-7	3	3	378	306	7	2	-7	1135	1118	-8	2	1	208	152	
-6	2	5	445	443	6	3	10	283	304	-7	3	-3	1054	1125	-7	2	7	405	451	8	2	2	364	493	
6	3	5	635	655	-6	3	10	486	490	7	3	-3	702	786	7	3	-7	175	193	* -8	2	-2	81	67	
-6	3	5	1351	1344	6	4	-10	621	637	7	4	3	648	664	* -7	3	-3	94	127	-8	2	-2	621	586	
* -6	3	5	94	295	6	5	-10	581	561	-7	4	3	1554	1666	-7	3	7	175	114	-8	3	2	378	543	
-6	3	5	364	303	-6	0	11	202	167	-7	4	-3	459	433	-7	4	7	973	1013	-8	3	2	1027	1161	
* -6	4	5	94	361	-6	1	11	594	602	7	4	-3	675	772	-7	4	-7	513	490	-8	3	-2	1310	1270	
6	4	-5	1243	1309	-6	1	11	1419	1484	-7	5	3	418	398	-7	5	7	202	138	8	3	-2	919	806	
-6	4	5	689	741	6	2	-11	554	527	-7	5	-3	472	405	7	5	-7	135	138	8	4	2	1000	1105	
6	5	-5	135	177	6	3	-11	919	920	-7	5	-3	391	290	7	6	-7	594	670	* -8	4	-2	81	56	
-6	5	5	581	609	7	0	0	1324	1239	-7	6	3	851	877	-7	6	7	743	702	-8	4	-2	824	820	
-6	5	-5	694	793	7	1	0	783	755	-7	6	-3	959	912	7	7	-7	243	254	-8	4	-4	243	188	
6	6	5	729	745	7	-1	0	2851	2656	-7	6	3	791	863	-7	7	7	345	385	-8	5	2	756	796	
-6	6	5	486	453	-7	2	0	662	622	-7	7	3	243	259	-7	0	8	1432	1483	-8	5	-2	608	562	
-6	6	5	432	491	7	2	0	283	259	-7	7	-3	1811	1702	7	1	-8	405	344	8	5	-2	513	464	
6	7	-5	513	517	-7	3	0	1851	1812	7	7	-3	810	915	-7	1	8	851	872	8	6	-2	567	528	
-6	7	5	1108	1069	-7	3	0	892	853	-7	8	-3	216	135	-7	2	8	432	397	-8	6	2	135	105	
-6	7	-5	554	557	-7	4	0	1594	1546	-7	8	3	662	644	7	2	-8	1148	1105	-8	7	-2	1337	1267	
6	8	-5	635	643	7	5	0	743	718	-7	9	3	148	99	-7	3	8	702	728	-8	7	2	675	776	
-6	8	5	635	670	-7	5	0	2000	2028	-7	9	-3	459	512	7	3	-8	689	669	-8	8	2	270	283	
* -6	8	5	94	308	-7	6	0	297	270	7	0	-4	729	705	-7	4	8	1216	1269	-8	9	-2	513	449	
6	0	6	243	189	* -6	7	6	0	297	270	7	0	-4	243	209	7	4	-8	1986	2006	-8	9	2	256	265
6	0	-6	189	168	-7	7	0	716	703	-7	1	4	364	224	* -7	5	8	94	68	8	0	3	1419	1356	
-6	1	6	1229	1273	-7	8	0	1405	1420	-7	1	-4	2000	2013	7	5	-8	446	480	8	0	-3	540	497	
6	1	-6	189	130	-7	9	0	1743	1747	7	1	4	324	349	-7	6	8	283	306	8	0	-3	527	497	
-6	1	6	621	617	7	0	1	2000	2105	-7	1	4	878	888	7	6	-8	378	340	-8	1	3	202	234	
-6	1	-6	202	209	7	0	-1	2148	2121	7	2	4	581	625	* -7	0	9	187	187	8	1	3	1081	1003	
6	2	-2	1500	1517	-7	1	1	513	492	-7	2	4	243	280	7	1	-9	459	367	-8	1	1	1162	1159	
6	2	6	378	374	7	1	1	1567	1378	-7	2	4	2500	2566	-7	1	9	472	485	8	1	3	1351	1315	
-6	2	6	621	508	-7	1	-1	3419	3334	-7	2	4	3446	3523	-7	2	9	648	630	-8	2	3	729	645	
-6	2	6	1310	1336	-7	1	1	189	146	-7															

Table 2 (cont.)

H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL						
-9	4	0	1121	1049	-9	7	-4	567	520	-10	6	-1	783	956	10	0	-8	527	470	-11	7	5	756	709	-12	3	8	202	224	
-9	5	0	675	676	-9	8	4	216	201	-10	7	1	216	141	-10	1	8	310	356	11	0	-6	1013	1010*	13	-2	-1	94	207	
*-9	7	0	94	148	9	1	-5	1324	1372	-10	8	1	581	534	10	1	8	675	620	11	1	-6	351	348	-13	3	1	594	683	
-9	8	0	919	938	-9	1	5	1135	1228	-10	8	-1	1027	899	10	2	-8	810	822	-11	2	6	567	617*	13	-2	-2	94	153	
-9	9	0	513	493	-9	2	-5	1283	1223	10	0	-2	1000	998	10	3	-8	729	724	11	2	-6	283	261	-13	3	2	649	662	
9	0	1	1094	1128	-9	2	5	783	826	10	0	2	810	1023	-10	3	8	378	399	-11	3	6	527	536	-13	4	2	310	348	
9	0	-1	1081	1055	9	2	-5	175	145	10	-1	-2	919	867	-10	4	8	148	140	11	4	-6	432	382*	-13	5	2	94	150	
9	1	1	1040	1049	-9	3	5	2675	3025	10	1	2	81	116	10	4	-8	770	770	-11	4	6	689	683	13	0	-3	513	478	
9	-1	1	756	610	9	3	-5	500	578	10	-1	2	54	247	-10	5	8	324	297	-11	6	6	202	182	13	-1	-3	581	496	
9	-1	-1	540	493	9	3	-5	378	378	10	-2	2	527	710	-10	1	9	864	892	11	0	-7	1365	1387	-13	3	3	378	299	
9	-1	-10	1054	1027	-9	4	5	554	446	10	2	-2	1594	1486	10	1	9	743	733	11	1	-7	973	995	-13	4	3	216	271	
9	-2	1	810	758	-9	4	-5	919	924	-10	2	2	1459	1466*	10	2	-9	94	125	-11	2	7	148	102	-13	5	3	594	675	
9	2	1	1040	935	9	4	-5	648	635	10	3	-2	554	558	10	3	-9	810	794*	11	2	-7	94	62	13	-1	-4	364	348	
-9	2	1	635	652	-9	5	5	824	809	-10	3	2	837	809	-10	3	9	797	795	11	3	-7	432	426	13	-2	-4	1067	1012	
9	2	-1	797	726	9	5	-5	891	886	-10	3	-2	148	80	-10	4	9	378	348	-11	3	7	689	718	-13	3	4	1297	1361	
-9	3	1	473	491	-9	5	-5	297	321	-10	4	-2	1392	1376	10	0	-10	783	795	-11	4	7	608	613	-13	4	4	540	529	
-9	3	-1	837	758	9	6	-5	148	181	10	4	-2	540	485	-10	1	10	1229	1277	11	0	-8	202	202	-13	5	4	675	740	
9	3	1	270	214	-9	7	5	1392	1407	10	5	-2	554	516	-10	2	10	135	123	11	1	-8	202	115	13	-1	-5	500	484	
9	4	-1	378	365	-9	8	5	148	164	-10	5	2	1054	1060	11	0	0	986	929	-11	2	8	432	438	-13	2	5	445	447	
9	4	1	1288	1102	-9	1	6	1324	1418	-10	6	2	1094	1071	11	1	0	310	247	11	2	-8	594	554	-13	3	5	635	651	
-9	4	-1	608	589	9	1	-6	1202	1235	-10	6	-2	878	880	11	-1	0	864	783	11	3	-8	432	459	-13	4	5	716	690	
-9	5	1	905	859	-9	2	6	891	928	-10	7	2	337	298	11	2	0	405	378*	-11	3	8	94	128	13	0	-6	527	562	
-9	5	-1	837	940	9	2	-6	459	453*	-10	7	-2	94	140	11	-2	0	662	684	11	0	-9	851	828	13	-1	-6	851	853	
-9	6	-1	567	522	-9	3	6	1946	1992	-10	8	2	310	238	-11	3	0	540	560	-11	1	9	283	326	-13	2	6	810	784	
-9	6	1	540	624	9	3	-6	135	155	10	0	3	405	526	-11	4	0	1459	1360	11	1	-9	608	566	-13	3	6	297	210	
-9	6	1	635	573	-9	4	6	135	131	10	0	-3	919	861	-11	5	0	459	389	-11	2	9	243	159	-13	4	6	216	142	
-9	7	1	702	781	9	4	-6	743	818	10	-1	3	973	978	-11	6	0	851	850	-11	3	9	473	437	13	-1	-7	283	253	
-9	8	1	148	111	-9	5	6	486	501	10	-1	3	986	1113	11	0	-1	256	135	12	-1	0	1040	1137	-8	7	-5	797	866	
*-9	8	-1	94	79	-9	6	459	432	10	2	-3	1716	1644	11	0	1	905	946	12	-2	0	216	164							
-9	9	1	1419	1327	9	6	-6	148	101	10	-2	3	635	721	11	-1	1	756	759	-12	3	0	783	772						
-9	9	-1	486	657	-9	7	6	689	734	-10	2	3	783	795	11	-1	1	864	768	-12	4	0	1243	1407						
-9	0	-2	283	292	9	0	-7	189	136	10	3	-3	1189	1113	11	-1	1	364	327	-12	5	0	1756	1742						
9	0	-2	1973	2151	-9	1	7	851	905	10	-3	3	351	374	11	1	1	540	410	-12	6	0	270	259						
-9	1	-1	2216	2137	9	1	-7	229	213	-10	3	3	270	275	11	-2	-1	527	472	12	0	-1	148	112						
9	1	-2	486	437	9	2	-7	797	798	10	4	-3	500	444	11	-2	1	648	590	12	-1	-1	743	772						
9	1	-2	851	964	-9	2	7	621	651	-10	4	-3	378	388	11	2	1	865	832	12	1	-1	648	638						
9	-1	2	310	543	-9	3	7	621	641	-10	4	3	310	321	-11	3	-1	432	437	12	-2	1	162	130						
9	-2	2	121	218	9	3	-7	1067	1038	-10	5	3	675	732	-11	3	1	1135	1086	12	-2	-1	216	187						
-9	2	2	1459	1359	-9	4	7	729	612	-10	5	-3	486	478	11	3	-1	310	278	-12	3	1	668	787						
9	2	-2	1486	1425	9	4	-7	648	622	10	5	-3	256	237	-11	4	-1	2013	1995	-12	3	1	756	785						
9	2	-2	1000	1093*	9	5	-7	94	67	-10	6	-3	702	706	-11	4	1	608	509	-12	4	-1	270	284						
-9	3	2	94	54	-9	5	7	608	616	-10	6	-3	459	490	-11	5	1	919	853	-12	4	1	94	120						
-9	3	2	324	307	-9	6	7	783	810	-10	7	3	594	557	-11	5	1	1621	1523	-12	5	1	973	899						
-9	3	-2	1013	934	9	0	-8	405	348	-10	8	3	337	379	-11	6	1	513	487	-12	5	-1	973	890						
-9	4	-2	689	668	-9	1	8	513	481	10	0	-4	135	138	-11	6	1	270	283	-12	0	-2	770	725						
-9	4	-2	1419	1419	-9	2	8	310	334	-10	1	-4	500	502	-11	7	1	148	121	-12	1	-2	216	211						
9	5	-2	148	228	9	2	-8	716	734	10	1	-4	1000	984*	11	0	2	94	63	12	1	-2	851	817						
-9	5	-2	1351	1393	9	3	-8	94	68	10	2	-4	581	584	11	0	-2	1067	974	12	2	-2	810	710						
-9	6	2	648	606	-9	3	8	486	508	-10	2	4	959	985	11	-1	2	408	486	-12	2	-2	256	218						
-9	7	2	2	905	852	9	4	-8	405	413	-10	3	-3	1202	1190	-11	2	2	500	407	-12	3	2	729	721					
-9	8	-2	932	999	9	5	-8	716	721	-10	4	-2	148	162	-11	2	2	256	242	-12	6	2	310	262						
*-9	8	2	94	131	9	0	-9	1121	1079	-10	4	4	1054	1092	-11	2	-2	621	591	12	0	-3	729	613						
-9	9	-3	297	289	9	1	-9	596	572	-10	4	-4	635	670	-11	3	2	202	162	12	1	-3	446	355						
-9	9	0	3	243	241	-9	2	9	378	390	10	5	-4	554	468	-11	3	-2	459	463	12	2	-3	1040	920					
-9	9	-1	3	1554	1466	-																								

related to it by a centre of inversion. Thus, for convenience of identification, Pd(1) for example, is related to Pd(1') by a centre of symmetry. A diagrammatic representation of the structure projected down the c^* and b^* axes is shown in Figs. 3 and 4 respectively. From these diagrams it becomes obvious that in this crystal structure the compound exists not as a monomer but as a hexamer consisting of six $\text{Pd}(\text{S}, \text{C}_3\text{H}_7)_2$ fragments. The basic unit of this large molecule consists of one palladium and four sulphur atoms arranged in an approximately square planar configuration. Six units of this type are linked to each other by the sharing of two sulphur atoms between every two adjacent palladium atoms. The sulphur atoms thus act as a bridge between two palladium atoms. The Pd—S—Pd bridge angles vary between 79.7 and 85.8°. The nature of the sharing of sulphur atoms can be seen in Fig. 2(a). In the resulting structure then, there are two six-membered sulphur rings which are S(1), S(3), S(5'), S(6'), S(2'), S(4) and S(1'), S(3'), S(5), S(6), S(2), S(4'). These are packed approximately above each other and one

ring is related to the other by a centre of inversion. The average distance between the two rings is 3.09 Å. There is also a ring of six palladium atoms which is

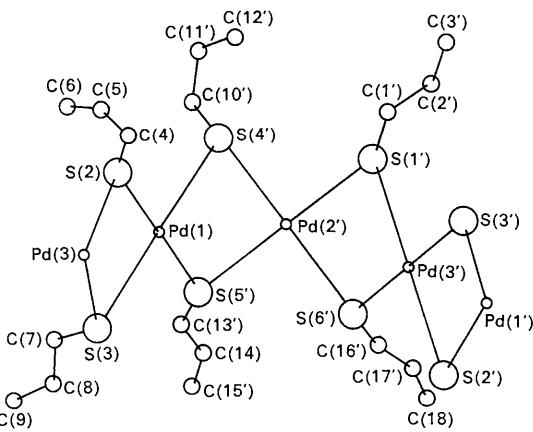


Fig. 1. Schematic diagram of a part of the molecule of palladium n-propyl mercaptide.

Table 3. Dimensions of the molecule

	Bond length (Å)	σ (Å)		Angles (°)	σ (°)
Pd(1)—S(2)	2.305	0.004	S(5)—Pd(1)—S(4)	82.0	0.2
Pd(1)—S(3)	2.399	0.004	S(5)—Pd(1)—S(3)	98.2	0.2
Pd(1)—S(4)	2.413	0.004	S(5)—Pd(1)—S(2)	177.8	0.2
Pd(1)—S(5)	2.314	0.005	S(4)—Pd(1)—S(3)	176.6	0.2
			S(4)—Pd(1)—S(2)	97.6	0.2
			S(3)—Pd(1)—S(2)	82.1	0.2
Pd(2)—S(1)	2.450	0.005	S(6)—Pd(2)—S(5)	98.7	0.2
Pd(2)—S(4)	2.358	0.004	S(6)—Pd(2)—S(4)	174.6	0.2
Pd(2)—S(5)	2.454	0.004	S(6)—Pd(2)—S(1)	80.3	0.2
Pd(2)—S(6)	2.331	0.005	S(5)—Pd(2)—S(4)	80.3	0.2
			S(5)—Pd(2)—S(1)	177.9	0.2
			S(4)—Pd(2)—S(1)	100.5	0.2
Pd(3)—S(1)	2.282	0.004	S(6)—Pd(3)—S(3)	174.0	0.2
Pd(3)—S(2)	2.288	0.004	S(6)—Pd(3)—S(2)	94.7	0.2
Pd(3)—S(3)	2.303	0.005	S(6)—Pd(3)—S(1)	84.5	0.2
Pd(3)—S(6)	2.308	0.005	S(3)—Pd(3)—S(2)	84.6	0.2
			S(3)—Pd(3)—S(1)	96.0	0.2
			S(2)—Pd(3)—S(1)	177.4	0.2
S(1)—C(1)	1.82	0.02	C(1)—S(1)—Pd(3)	113.3	0.7
C(1)—C(2)	1.60	0.03	C(1)—S(1)—Pd(2)	110	0.7
C(2)—C(3)	1.46	0.04	C(2)—C(1)—S(1)	110	1.4
			C(3)—C(2)—C(1)	110	2.0
S(2)—C(4)	1.92	0.02	C(4)—S(2)—Pd(1)	104.9	0.6
C(4)—C(5)	1.62	0.03	C(4)—S(2)—Pd(3)	110.5	0.6
C(5)—C(6)	1.46	0.04	C(5)—C(4)—S(2)	104	1.4
			C(4)—C(5)—C(6)	115	2.2
S(3)—C(7)	1.89	0.05	C(7)—S(3)—Pd(3)	105	1.6
C(7)—C(8)	1.52	0.03	C(7)—S(3)—Pd(1)	108	0.6
C(8)—C(9)	1.59	0.04	C(8)—C(7)—S(3)	116	1.7
			C(7)—C(8)—C(9)	113	2.3
S(4)—C(10)	1.81	0.02	C(10)—S(4)—Pd(1)	99	0.6
C(10)—C(11)	1.58	0.03	C(10)—S(4)—Pd(2)	107	0.6
C(11)—C(12)	1.48	0.04	C(11)—C(10)—S(4)	114	1.4
			C(10)—C(11)—C(12)	110	201
S(5)—C(13)	1.84	0.02	C(13)—S(5)—Pd(1)	112	0.6
C(13)—C(14)	1.43	0.03	C(13)—S(5)—Pd(2)	110	0.7
C(14)—C(15)	1.50	0.03	C(14)—C(13)—S(5)	113	1.4
			C(13)—C(14)—C(15)	111	2.0
S(6)—C(16)	1.88	0.02	C(16)—S(6)—Pd(3)	100	0.7
C(16)—C(17)	1.47	0.04	C(16)—S(6)—Pd(2)	105	0.7
C(17)—C(18)	1.54	0.04	C(17)—C(16)—S(6)	115	1.8
			C(16)—C(17)—C(18)	112	2.8

sandwiched between the two sulphur rings in an anti-prismatic configuration. Mean planes passing through the palladium and sulphur atoms were calculated to see if these rings were planar. From the results of these calculations it can be seen that (see Table 6) the rings are not planar but puckered. The dimensions of the rings are shown in Figs. 2(a) and 2(b). The sulphur-sulphur bonds in the sulphur ring vary between 3.38–3.70 Å and the S-S-S angles between 114.1°–124.2°. In the palladium ring the palladium distances vary between 3.014 and 3.235 Å and the angles between 116.3° and 123.7°.

The n-propyl groups to which sulphur is bonded are located outside the framework of palladium and sulphur atoms. Since the carbon-sulphur bond is a single bond, the n-propyl groups can rotate around this bond

and orient themselves to sterically suitable positions in the crystal structure. The geometry of the molecule of $(\text{Pd } n\text{-propyl mercaptide})_6$ is very interesting. The molecule has a cage structure and has a large hole of 7 Å diameter within it.

The Pd-S distances vary between 2.282 and 2.454 Å. Mean planes were calculated to test the planarity of the PdS_4 units. The deviations of a number of atoms from the mean planes are significant in all the three independent PdS_4 units. Of the six different carbon-sulphur bonds, three are normal, the lengths being about 1.82 Å. The remaining three bonds have somewhat larger value (see Table 3). The agreement between the carbon-carbon distances in the six independent n-propyl groups is not very satisfactory. But considering that the carbon atoms were located in the presence of heavy palladium and sulphur atoms, little significance need be attached to these deviations.

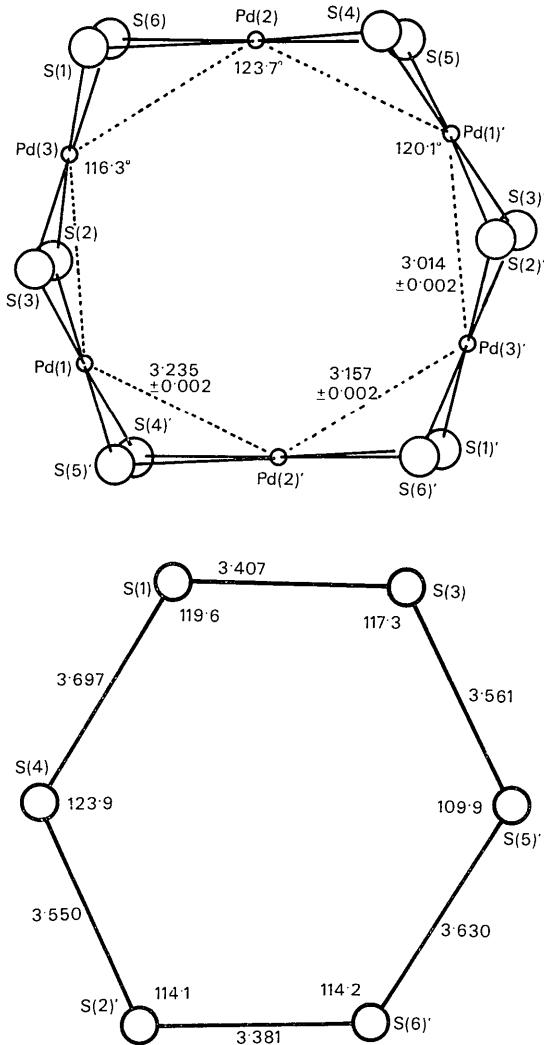


Fig. 2. A diagram showing (a) the dimensions of the ring of six palladium atoms and the sharing of sulphur atoms between them. (b) The dimensions of the ring of six sulphur atoms.

Table 4. Intermolecular approaches

	Lenght (Å)	σ (Å)
C(6')—C(6)	4.39	0.04
C(12')—C(18)	4.02	0.04
C(9)—C(18)	5.10	0.04
C(9)—C(9')	4.39	0.04
C(9)—C(18')	4.27	0.04
C(6)—C(12)	4.66	0.04
C(6)—C(18')	4.10	0.04
C(9)—C(15)	5.33	0.04
C(9)—C(12)	4.55	0.04
C(7)—C(15)	3.81	0.04
C(16)—C(18)	4.53	0.04
C(6)—C(9)	4.19	0.04

The manner in which the various groups are packed in the crystal structure can be seen in Figs. 3 and 4. The various intermolecular close approaches are listed in Table 4. The short intermolecular approaches are those between the terminal methyl groups of the n-propyl chains, and their values vary between 3.81 and 5.33 Å. None of their values are smaller than the van der Waals diameter of the methyl group.

It is a pleasure to thank Professor David Harker for his interest and encouragement. Thanks are also due to Dr Ray Davis for his assistance in the use of the crystallographic programs, to Mr C. T. Lu for his help in the use of the G.E. XRD-6 diffractometer and the staff of the Computing Centre of the Roswell Park Memorial Institute for their assistance. This work was supported by grants NSF-6B-4056 from the National

Table 5. Sulphur bridge angles

Pd(1)—S(2)—Pd(3)	$82.0^\circ \pm 0.1^\circ$
Pd(1)—S(3)—Pd(3)	79.7 ± 0.1
Pd(3)—S(1)—Pd(2)	83.6 ± 0.1
Pd(3)—S(6)—Pd(2)	85.8 ± 0.1
Pd(2)—S(4)—Pd(1')	85.4 ± 0.1
Pd(2)—S(5)—Pd(1')	85.4 ± 0.1

Science Foundation, NIH-A-3942 from the National Institutes of Health and Roswell Park Buildings Research Grant.

Table 6. Least-squares planes and the out-of-plane distances of the plane determining atoms

Plane number 1. This plane is defined by 5 atoms. The equation of the plane is $0.1524P + 0.9550Q - 0.2546R = 0.242 \text{ \AA}$. The standard deviation of the distances from the plane is 0.613 \AA . Out-of-plane distances of the plane-determining atoms are:

Pd(1)	-1.130 \AA
S(3)	0.495
S(4')	0.086
S(5')	-0.041
S(2)	0.590

Plane number 2. This plane is defined by 5 atoms. The equation of the plane is $0.1168P + 0.9494Q + 0.2916R = 3.738 \text{ \AA}$. The standard deviation of the distances from this plane is 0.306 \AA . Out-of-plane distances of the plane-determining atoms are:

Pd(2)	-0.384 \AA
S(1)	0.067
S(5)	-0.330
S(4)	0.334
S(6)	0.313

Plane number 3. This plane is defined by 5 atoms. The equation of the plane is $0.5168P + 0.8045Q + 0.2929R = 2.971 \text{ \AA}$. The standard deviation of the distances from the plane is 0.2197 \AA . Out-of-plane distances of the plane-determining atoms are:

Pd(3)	0.141 \AA
S(1)	0.096
S(2)	0.095
S(3)	-0.438
S(6)	0.107

Plane number 4. This plane is defined by 6 atoms. The equation of the plane is $-0.4961P + 0.8180Q - 0.2912R = 0.000 \text{ \AA}$. The standard deviation of the distances from the plane is 1.160 \AA . Out-of-plane distances of the plane-determining atoms are:

Pd(1)	-1.683
Pd(2)	0.242
Pd(3)	0.884
Pd(1')	1.683
Pd(2')	-0.242
Pd(3')	-0.884

Plane number 5. This plane is defined by 6 atoms. The equation of the plane is $0.1147P + 0.9698Q - 0.2154R = -0.771 \text{ \AA}$. The standard deviation of the distances from the plane is 1.342 \AA . Out-of-plane distances of the plane-determining atoms are:

S(2)	1.605 \AA
S(5)	0.471
S(6)	-0.377
S(1')	-2.604
S(3')	-0.129
S(4')	1.035

Angles between the planes

Plane A	Plane B	Angle
1	3	140.6°
2	3	155.4°
1	4	148.2°

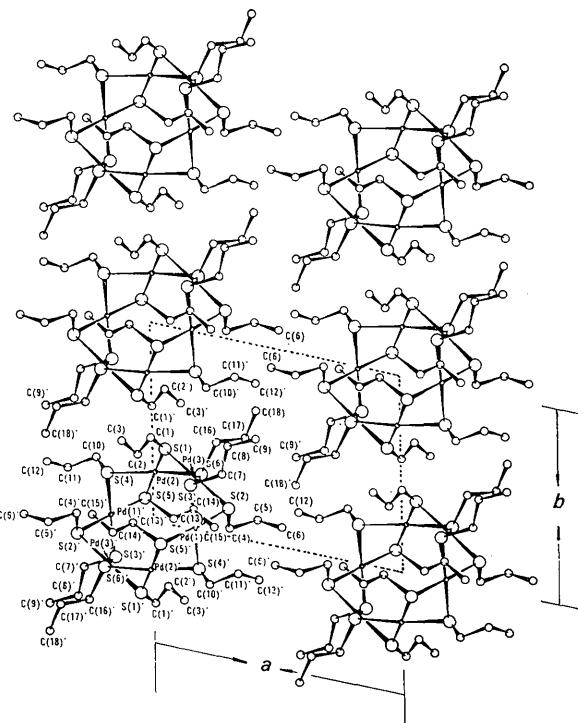


Fig. 3. A diagrammatic representation of the structure projected down the c^* axis.

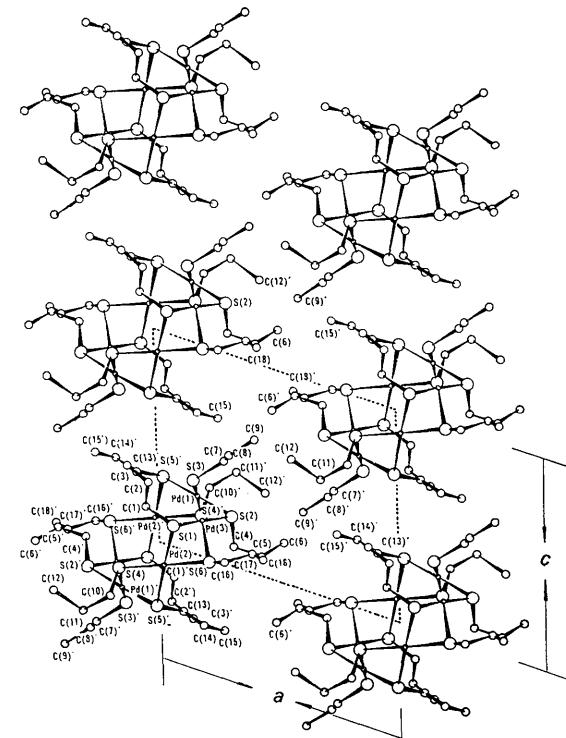


Fig. 4. A diagrammatic representation of the structure projected down the b^* axis.

Table 7. Anisotropic thermal vibration parameters
 The B_{ij} are coefficients in the temperature factor expression:
 $\exp [-(B_{11}h^2 + B_{22}k^2 + B_{33}l^2 + B_{12}hk + B_{13}hl + B_{23}kl)]$
 $U_{11} = B_{11}/2\pi^2a^{*2}; U_{12} = B_{12}/4\pi^2a^{*}b^{*}; \text{etc.}$

	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Pd(1)	0.0399	0.0489	0.0496	0.000	-0.0015	-0.006
Pd(2)	0.0451	0.0486	0.0503	0.0000	-0.0047	0.0019
Pd(3)	0.0402	0.0489	0.0481	0.0003	-0.0026	-0.0009
S(1)	0.0494	0.0474	0.0520	0.0064	-0.0062	-0.0031
S(2)	0.0348	0.0516	0.0523	0.0005	0.0010	0.0001
S(3)	0.0457	0.0506	0.0563	0.0047	0.0109	0.0026
S(4)	0.0466	0.0551	0.0494	0.0029	-0.0078	-0.0035
S(5)	0.0444	0.0560	0.0473	-0.0020	-0.0003	0.0079
S(6)	0.0459	0.0599	0.0503	-0.0019	0.0000	0.0046
C(1)	0.0502	0.0841	0.0785	0.0046	0.0174	0.0017
C(2)	0.0683	0.0857	0.0651	0.0120	0.0176	-0.0072
C(3)	0.1049	0.1377	0.1272	0.0057	0.0290	0.0041
C(4)	0.0467	0.0890	0.0536	0.0019	-0.0019	-0.0130
C(5)	0.0879	0.1300	0.0895	-0.0247	0.0091	-0.0163
C(6)	0.0638	0.1927	0.1152	0.0340	-0.0090	0.0099
C(7)	0.0776	0.0691	0.0627	-0.0059	0.0150	-0.0005
C(8)	0.2111	0.0819	0.1163	0.0313	-0.0545	-0.0428
C(9)	0.2451	0.1225	0.0999	0.0203	-0.0841	-0.0454
C(10)	0.0747	0.0621	0.0629	-0.0063	-0.0234	0.0154
C(11)	0.0787	0.0839	0.0648	0.0032	-0.0334	-0.0021
C(12)	0.00469	0.1519	0.1819	0.0334	-0.0334	-0.0142
C(13)	0.0516	0.0604	0.0842	0.0172	0.0172	0.0082
C(14)	0.0388	0.0991	0.0570	-0.0072	-0.0015	0.0136
C(15)	0.0777	0.1553	0.1021	0.0184	-0.0138	0.0190
C(16)	0.0709	0.0874	0.0998	-0.0600	-0.0099	0.0113
C(17)	0.0840	0.0939	0.2147	-0.0096	0.0059	0.0131
C(18)	0.1264	0.1119	0.2526	-0.0426	-0.0343	-0.0337

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The Crystal and Molecular Structure of 4,4'-Diamino-3,3'-dichlorobiphenyl

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(Received 19 December 1967)

4,4'-Diamino-3,3'-dichlorobiphenyl, $C_{12}H_{10}N_2Cl_2$, crystallizes with cell dimensions $a=12.5$, $b=3.85$, $c=23.7$ Å, $\beta=108^\circ$, space group $P2_1/c$ and $Z=4$. The structure has been determined from three-dimensional X-ray data and refined by the minimum residual method, with isotropic temperature factors for individual atoms, to give a final R index of 14%. The two phenyl rings are not coplanar, but are twisted around the linkage between them, C(1)-C(1'), so as to be mutually inclined at an angle of 21°; in addition, each phenyl ring is bent through a small angle (approximately 2.2° and 3.4° respectively) away from the line C(1)-C(1'). The length of the bond C(1)-C(1') is 1.515 ± 0.024 Å.

Introduction and experimental

A preliminary examination of one projection of the structure of 4,4'-diamino-3,3'-dichlorobiphenyl by

Toussaint (1948) gave an electron density map with spurious symmetry in addition to the true symmetry of the actual structure. The map suggested that the molecules are probably planar, or very nearly so, with chlorine atoms in the *trans* positions. We have now completed the analysis of the structure, using three-dimensional X-ray data kindly supplied by Dr Toussaint, and find that the phenyl rings are mutually inclined at an angle of 21°.

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